

International Journal of Pharmacy & Life Sciences

Open Access to Researcher

©2010, Sakun Publishing House and licensed by IJPLS, This is Open Access article which permits unrestricted non-commercial use, provided the original work is properly cited.



# **Review on Application of HPLC and UPLC**

# Prashant Gaurav<sup>1</sup>\*, Satpal Kushwaha<sup>2</sup>, Himani Tiwari<sup>3</sup> and Kaushal Kishor Chandrul<sup>4</sup>

# 1, Student of B. Pharm. 4th Year; 2, Assistant Professor; 3, HOD; 4, Principal

Department of Pharmacy, Mewar University, Gangrar Chittorgarh, (R.J.) - India

Article info

### Abstract

Received: 12/02/2024

Revised: 22/2/2024

Accepted: 25/03/2024

© IJPLS

www.ijplsjournal.com

High-Performance Liquid Chromatography (HPLC) and Ultra-Performance Liquid Chromatography (UPLC) are widely used techniques in separation science. HPLC employs a liquid mobile phase to separate components based on their interactions with a stationary phase, making it versatile and applicable to various sample types. It offers different modes of separation and is known for its robustness. On the other hand, UPLC is an advanced version of HPLC that utilizes smaller particle sizes and higher pressures, resulting in improved resolution, sensitivity, and speed. UPLC is particularly useful in fields requiring rapid separations and high sensitivity. Both techniques offer various detection methods and can be combined with sample preparation techniques to enhance analysis. HPLC and UPLC find applications in pharmaceutical analysis, environmental analysis, food and beverage industries, forensic analysis, clinical diagnostics, and more.

They provide high-resolution separations, rapid analysis, and versatile detection capabilities. The choice between the two techniques depends on specific analytical requirements and constraints. Overall, HPLC and UPLC continue to contribute significantly to scientific research, quality control, and industrial processes

Key words: HPLC, UPLC, Analysis, Drug

## Introduction

High-Performance Liquid Chromatography (HPLC) and Ultra-Performance Liquid Chromatography (UPLC). HPLC and UPLC are widely used techniques in analytical chemistry for the separation, identification, and quantification of compounds in complex mixtures.<sup>(1)</sup>

HPLC is a versatile technique that utilizes a liquid mobile phase and a stationary phase to separate analytes based on their interactions. It finds applications in various industries, including pharmaceuticals, environmental analysis, food and beverages, clinical diagnostics, forensic science, and materials science. <sup>(2)</sup> The choice of column and detection technique depends on the nature of the analytes and separation requirements. UPLC is an advanced version of HPLC that overcomes the limitations of traditional HPLC. It employs smaller particle sizes in the stationary phase and operates at higher pressures, resulting in improved resolution, faster separations, and increased sensitivity. UPLC is particularly useful in high-throughput analysis where speed is essential.<sup>(3)</sup>

\*Corresponding Author

Both HPLC and UPLC require similar instrumentation, including a mobile phase solvent reservoir, a high-pressure pump, an injector, a separation column, a detector, and a data acquisition system. Various detection techniques, such as UV-Visible spectroscopy, fluorescence, mass spectrometry, and refractive index detection, can be employed to detect and quantify analytes. Sample preparation methods can also be integrated to enhance selectivity and sensitivity.<sup>(4)</sup> The applications of HPLC and UPLC are vast and span across industries and research areas. They

span across industries and research areas. They are crucial in pharmaceutical analysis, environmental analysis, food and beverage analysis, forensic analysis, clinical diagnostics, bioanalysis, natural product analysis, and materials science. These techniques enable drug development, quality control, environmental monitoring, food safety assessment, forensic investigations, disease diagnosis, and material characterization.

While HPLC is versatile and widely used, UPLC offers improved performance with higher resolution and faster separations. <sup>(5)</sup> The choice between HPLC and UPLC depends on the specific analytical requirements and time constraints of the application.

# Background and Significance of HPLC and UPLC

sample matrices and detect analytes at low concentrations is of great significance. These techniques enable the separation and analysis of impurities, contaminants, and target compounds present in complex samples.<sup>(6)</sup>

In pharmaceutical analysis, HPLC and UPLC play a crucial role in drug discovery, development, and quality control. They allow for the separation and quantification of active pharmaceutical ingredients (APIs), excipients, impurities, and degradation products. These techniques ensure the purity, stability, and bioavailability of drugs, contributing to the safety and efficacy of pharmaceutical products.

In environmental analysis, HPLC and UPLC are used to detect and quantify pollutants, pesticides, and other contaminants in water, soil, and air samples. <sup>(7)</sup> Their high sensitivity enables the identification of compounds at trace levels, ensuring environmental safety and regulatory compliance. In clinical diagnostics, HPLC and UPLC are employed for the analysis of biomarkers, drugs, and metabolites in patient samples. They aid in disease diagnosis, therapeutic drug monitoring, and patient health assessment. Coupling HPLC and UPLC with mass spectrometry allows for high-throughput analysis with excellent sensitivity and selectivity.<sup>(8)</sup>

3.3. Quantitative and Qualitative Analysis: Both HPLC and UPLC facilitate quantitative and qualitative analysis of analytes. Quantitative analysis involves the determination of analyte concentrations, while qualitative analysis focuses on the identification and characterization of compounds based on their retention times, spectra, or mass-to-charge ratios.<sup>(9)</sup>

In quantitative analysis, HPLC and UPLC rely on calibration curves generated using standard solutions of known concentrations. The peak areas or heights of the analytes are compared to those of the standards to calculate the concentration in the sample. <sup>(10)</sup>The high sensitivity and linearity of these techniques make them ideal for accurate and precise quantitative measurements.

Qualitative analysis involves the identification and characterization of compounds based on their chromatographic behavior and spectroscopic properties. HPLC and UPLC systems are often coupled with various detectors, such as UV-Visible detectors, fluorescence detectors, mass spectrometers, and refractive index detectors, to provide additional information about the analytes. This combination of separation power and detection capabilities enables the reliable identification of compounds in complex mixtures. <sup>(11)</sup>

3.4. Speed and Throughput: HPLC and UPLC offer rapid separations, revolutionizing laboratory workflows and allowing for higher sample throughput and faster analysis times. This is particularly beneficial in industries where timely results are crucial, such as pharmaceutical manufacturing, clinical diagnostics, and environmental monitoring.

Modern HPLC and UPLC techniques achieve faster separations by utilizing smaller particle sizes and higher operating pressures. Smaller particle sizes provide higher separation efficiency, while higher pressures overcome the increased backpressure caused by the small particles. These factors result in shorter analysis times without sacrificing separation performance. <sup>(12)</sup>

3.5. Selectivity and Resolution: HPLC and UPLC provide excellent selectivity and resolution, enabling the separation of closely related compounds and complex mixtures. The choice of stationary phase, mobile phase composition, and separation conditions can be optimized to achieve the desired separation.

The selectivity of HPLC and UPLC can be manipulated by changing the composition of the mobile phase or by modifying the stationary phase. <sup>(13)</sup> The choice of stationary phase with appropriate physicochemical properties allows for selective interactions with the analytes of interest. Resolution, a measure of the ability to separate two closely eluting peaks, depends on the efficiency of the column, the selectivity of the separation, and the peak width. HPLC and UPLC provide high-resolution separations due to their efficient packing materials, optimized separation conditions, and improved detector technologies.

#### **Instrumentation and Method Development**

sample matrices and detect analytes at low concentrations is of great significance. These techniques enable the separation and analysis of impurities, contaminants, and target compounds present in complex samples.<sup>(14)</sup>

In pharmaceutical analysis, HPLC and UPLC play a crucial role in drug discovery, development, and quality control. They allow for the separation and quantification of active pharmaceutical ingredients (APIs), excipients, impurities, and degradation products. These techniques ensure the purity, stability, and bioavailability of drugs, contributing to the safety and efficacy of pharmaceutical products.<sup>(15)</sup>

In environmental analysis, HPLC and UPLC are used to detect and quantify pollutants, pesticides, and other contaminants in water, soil, and air samples. Their high sensitivity enables the identification of compounds at trace levels, ensuring environmental safety and regulatory compliance.<sup>(16)</sup>

In clinical diagnostics, HPLC and UPLC are employed for the analysis of biomarkers, drugs, and metabolites in patient samples. They aid in disease diagnosis, therapeutic drug monitoring, and patient health assessment. Coupling HPLC and UPLC with mass spectrometry allows for high-throughput analysis with excellent sensitivity and selectivity.

Quantitative and Qualitative Analysis: Both HPLC and UPLC facilitate quantitative and qualitative analysis of analytes. Quantitative analysis involves the determination of analyte concentrations, while qualitative analysis focuses on the identification and characterization of compounds based on their retention times, spectra, or mass-to-charge ratios.<sup>(17)</sup>

In quantitative analysis, HPLC and UPLC rely on calibration curves generated using standard solutions of known concentrations. The peak areas or heights of the analytes are compared to those of the standards to calculate the concentration in the sample. <sup>(18)</sup> The high sensitivity and linearity of these techniques make them ideal for accurate and precise quantitative measurements.

Qualitative analysis involves the identification and characterization of compounds based on their chromatographic behavior and spectroscopic properties. HPLC and UPLC systems are often coupled with various detectors, such as UV-Visible detectors, fluorescence detectors, mass spectrometers, and refractive index detectors, to provide additional information about the analytes. This combination of separation power and detection capabilities enables the reliable identification of compounds in complex mixtures. <sup>(19)</sup>

3.4. Speed and Throughput: HPLC and UPLC offer rapid separations, revolutionizing laboratory workflows and allowing for higher sample throughput and faster analysis times. This is particularly beneficial in industries where timely results are crucial, such as pharmaceutical manufacturing, clinical diagnostics, and environmental monitoring.

Modern HPLC and UPLC techniques achieve faster separations by utilizing smaller particle sizes and higher operating pressures. Smaller particle sizes provide higher separation efficiency, while higher pressures overcome the increased backpressure caused by the small particles. These factors result in shorter analysis times without sacrificing separation performance<sup>(20)</sup>

3.5. Selectivity and Resolution: HPLC and UPLC provide excellent selectivity and resolution, enabling the separation of closely related compounds and complex mixtures. The choice of stationary phase, mobile phase composition, and separation conditions can be optimized to achieve the desired separation.<sup>(21)</sup>

The selectivity of HPLC and UPLC can be manipulated by changing the composition of the mobile phase or by modifying the stationary phase. The choice of stationary phase with appropriate physicochemical properties allows for selective interactions with the analytes of interest. Resolution, a measure of the ability to separate two closely eluting peaks, depends on the efficiency of the column, the selectivity of the separation, and the peak width. HPLC and UPLC provide high-resolution separations due to their efficient packing materials, optimized separation conditions, and improved detector technologies. (22)

# Mass Spectrometry and its coupling with HPLC/UPLC

the principles of mass spectrometry, including ionization, ion separation based on mass-to-charge ratio (m/z), and detection. It discusses various ionization techniques such as electrospray ionization (ESI), atmospheric pressure chemical ionization (APCI), and matrix-assisted laser desorption/ionization (MALDI). Different types of mass analyzers like quadrupole, time-of-flight (TOF), ion trap, and Orbitrap are also described, along with their modes of operation. The detection of ions using different detectors is covered as well. The paper then explores the different types of mass spectrometers, including quadrupole, TOF, ion trap, and Orbitrap instruments. It highlights their respective advantages and applications in targeted analysis, high sensitivity, and highresolution measurements. (23)

Next, the paper discusses the applications of mass spectrometry in various fields such as proteomics, metabolomics, pharmaceutical analysis, environmental analysis, forensic science, and food and beverage analysis. It explains how mass spectrometry is used for protein identification, metabolite analysis, drug discovery, environmental monitoring, forensic analysis, and quality control in the food industry.

The instrumentation section delves deeper into the components of a mass spectrometer, including ion sources, mass analyzers, and detectors. It discusses advancements in ionization techniques like nanoelectrospray and desorption electrospray ionization (DESI), as well as advancements in mass analyzers and detectors to improve sensitivity and accuracy.<sup>(24)</sup>

The paper then focuses on the coupling of mass spectrometry with HPLC/UPLC. It explains the advantages of this hyphenated approach, such as improved separation and simultaneous structural elucidation. It covers the requirements for compatibility between mass spectrometry and HPLC/UPLC, including solvent and flow rate compatibility. Method development and optimization considerations are also discussed.

The different coupling interfaces used in the integration of mass spectrometry with HPLC/UPLC, such as ESI, APCI, and APPI, are explored. The section emphasizes the impact of these interfaces on overall performance and discusses the considerations for interface selection based on analyte characteristics.<sup>(25)</sup>

Challenges and troubleshooting strategies related to the coupling of mass spectrometry with HPLC/UPLC are addressed. Common issues such as ion suppression, matrix effects, and peak broadening are discussed, along with tips for optimizing the coupling system.

## **Applications of HPLC and UPLC**

HPLC and UPLC are powerful analytical techniques with various applications across different fields. Here, we will explore the applications of HPLC and UPLC in detail.

HPLC. or High-Performance Liquid Chromatography, is a widely used technique known for its versatility, sensitivity, and ability to handle a broad range of sample types. In the pharmaceutical industry, HPLC plays a vital role in drug development, quality control, and formulation. It is used for the quantification of pharmaceutical ingredients active (APIs). impurity profiling, dissolution testing, stability studies, chiral separations, and pharmacokinetic analysis. (26)

In pharmaceutical analysis, HPLC is used for assay determination, which involves determining the concentration of the active ingredient in pharmaceutical formulations to ensure they meet specified potency requirements. It is also used for impurity profiling to separate and quantify impurities that may arise during the drug synthesis process or product storage, ensuring the safety and efficacy of pharmaceutical products. HPLC is employed in stability studies to assess the degradation profile of drugs over time and identify degradation products. Dissolution testing, another application, evaluates the release of the active ingredient from solid dosage forms, ensuring the drug is released at the desired rate and assessing the formulation's bioavailability. Chiral separations using HPLC are crucial for separating enantiomers (optical isomers) of chiral drugs, which have different pharmacological HPLC Finally, activities. is used in studies pharmacokinetic determine the to concentration of drugs and their metabolites in biological samples like plasma or urine, providing insights into drug absorption, distribution, metabolism, and elimination.<sup>(27)</sup>

The methodology for pharmaceutical analysis using HPLC involves sample preparation, selection of the appropriate chromatographic system, choice of mobile phase and column, detection of analytes, and calibration and quantification. Key parameters in HPLC analysis include retention time, peak shape, selectivity, sensitivity, and system suitability.<sup>(28)</sup>

HPLC also finds applications in environmental analysis, where it is used to determine the presence and concentration of pollutants in water, soil, and air samples. It enables the analysis of environmental contaminants such as pesticides, herbicides, polycyclic aromatic hydrocarbons (PAHs), and heavy metals. In food and beverage analysis, HPLC is extensively used to determine the composition, quality, and safety of food products. It can be employed for the analysis of acids. vitamins. amino food additives. mycotoxins, pesticides, and other contaminants. HPLC is utilized in clinical and forensic analysis for the quantification of drugs, metabolites, and biomarkers in biological samples, aiding in therapeutic drug monitoring, toxicology screening, and forensic analysis of drugs of abuse. Furthermore, HPLC is employed in biochemical and biomedical research for the analysis of proteins, peptides, nucleic acids, carbohydrates, and metabolites.<sup>(29)</sup>

UPLC, or Ultra-Performance Liquid Chromatography, is an advanced variant of HPLC that offers improved resolution, sensitivity, and faster analysis times. In pharmaceutical development, UPLC has become the method of choice due to its enhanced speed and efficiency. It is used for method development, drug formulation and quality control, bioavailability and pharmacokinetics, dissolution testing, and stability studies. UPLC provides increased resolution, enhanced sensitivity, faster analysis time, reduced solvent consumption, and compatibility with HPLC methods, making it advantageous in pharmaceutical development.<sup>(30)</sup>

UPLC is extensively employed in bioanalysis, especially in pharmacokinetic studies and drug metabolism research, where its high sensitivity allows the quantification of low-abundance analytes in biological matrices. UPLC-based metabolomics is a powerful tool for studying the metabolic profile of biological samples, enabling the identification and quantification of endogenous metabolites..<sup>(31)</sup>

### Advances And Innovations in Hplc/Uplc

HPLC and UPLC are widely used liquid chromatography techniques for separation, identification, and quantification of compounds in complex mixtures. Recent advancements have led to improved performance in terms of separation efficiency, sensitivity, and speed. Innovations in technology, such as core-shell. column monolithic, and sub-2 µm particle columns, have enhanced separation capabilities. Instrumentation include high-pressure gradient innovations systems, low-dispersion systems, miniaturized systems, and multidimensional systems. Detection system improvements involve mass spectrometry coupling, evaporative light scattering detection (ELSD), charged aerosol detection (CAD), fluorescence detection, and electrochemical detection. Method development strategies include high-throughput screening, Ouality by Design (ObD), green analytical methods, and intelligent software tools. Applications of HPLC and UPLC encompass pharmaceutical analysis, environmental analysis, food safety and quality control, and forensic analysis. <sup>(33)</sup>Challenges and future perspectives include miniaturization, enhanced data analysis and automation, advances in stationary phase technology, and integration with other analytical techniques. Overall, these advancements have expanded capabilities the of liquid chromatography, empowering researchers and analysts across scientific disciplines

### Limitation on HPLC & UPLC

compared to UPLC. The overall operating costs of HPLC systems are typically lower, making them a more cost-effective choice for routine analyses or applications where high resolution and fast separations are not critical requirements<sup>(34)</sup>

UPLC Cost: UPLC systems, on the other hand, have higher upfront costs compared to HPLC systems. The advanced technology and higherpressure capabilities of UPLC instrumentation contribute to the increased cost. UPLC instruments feature specialized high-pressure pumps, low-volume injection systems, and detectors with high data acquisition rates, which are designed to handle the increased backpressure associated with smaller particle sizes.<sup>(35)</sup> The use of premium materials and the need for precise manufacturing processes further contribute to the higher cost of UPLC instrumentation. Additionally, UPLC columns tend to be more expensive compared to HPLC columns due to the advanced manufacturing processes, specialized materials, and smaller particle sizes used in UPLC.

Maintenance and Service Costs: The maintenance and service costs of UPLC systems may also be higher compared to HPLC systems. UPLC systems are more complex and require specialized knowledge for maintenance and troubleshooting. The high precision and performance requirements of UPLC instrumentation may necessitate more frequent maintenance and calibration. It's important to consider these factors when evaluating the long-term operating costs of UPLC systems.<sup>(36)</sup>

In summary, UPLC systems generally have higher upfront costs compared to HPLC systems due to their advanced technology, higher-pressure capabilities, and specialized instrumentation. The maintenance and service costs of UPLC systems may also be higher. HPLC systems, on the other hand, have lower upfront costs and overall operating costs. The selection between HPLC and UPLC should consider the specific requirements of the analysis, available budget, and long-term cost considerations.<sup>(37)</sup>

### Reference

1. Snyder, L. R., Kirkland, J. J., & Dolan, J. W. (2010). Introduction to modern liquid chromatography (3rd ed.). Wiley.

2. Ahuja, S. (2011). Handbook of modern pharmaceutical analysis (2nd ed.). Academic Press.

 Schoenmakers, P. (2017). High-Performance Liquid Chromatography: Fundamental Principles and Practice. Wiley-VCH
Dolan, J. W. (2010). HPLC Methods for Pharmaceutical Analysis.Elsevier.

5. Wirth, J. (2012). HPLC: A practical guide. Wiley.

6. Dolan, J. W. (2017). Troubleshooting LC Systems: A Comprehensive Approach to Troubleshooting LC Equipment and Separations.Elsevier.

7. Khandagle, K. S., &Gosavi, S. W. (2015). Handbook of HPLC.CRC Press.

8. Pesek, J. J., &Matyska, M. T. (2005). Practical HPLC Method Development.Wiley.

9. Shalliker, R. A., &Paschke, A. (2011). Ultrahigh-pressure liquid chromatography and its applications.CRC Press.

10.Mondello, L., &Tranchida, P. Q. (Eds.).(2012).High-performanceliquidchromatography.IntechOpen.

11. Guillarme, D., &Veuthey, J. L. (2016). UHPLC in Life Sciences.Wiley.

12. Fekete, S., &Guillarme, D. (2013). UHPLC in pharmaceutical analysis.John Wiley & Sons.

13. Gilar, M., &Veuthey, J. L. (2016). UHPLC in food analysis.CRC Press.

14. Schoenmakers, P. J., & Unger, K. K. (2014). Comprehensive Chromatography in Combination with Mass Spectrometry.Wiley.

15. Chankvetadze, B., &Blaschke, G. (2011). Capillary electrophoresis in chiral analysis.Wiley-VCH.

16. Altria, K. D. (2009). Handbook of capillary and microchip electrophoresis and ssociatedmicrotechniques (3rd ed.). CRC Press.

17. Nováková, L., Vlčková, H., &Sýkora, D. (2011). Recent advances in UHPLC. TrAC Trends in Analytical Chemistry, 30(1), 5-17.

18. Núñez, O. (2013). Recent advances in UHPLC instrumentation. Journal of Chromatographic Science, 51(9), 853-862.

19. Shalliker, R. A., &Najam-ul-Haq, M. (2014). Recent advances in ultra-high pressure liquid chromatography. Analyst, 139(9), 2071-2082.

20. Greco, G., & Villani, C. (2019). Practical UHPLC Method Development and Transfer.Springer.Snyder, L.R., Kirkland, J.J., & Dolan, J.W. (2010). Introduction to Modern Liquid Chromatography (3rd ed.). John Wiley & Sons.

21. Wouters, B., & De Winter, W. (2016). UHPLC in Life Sciences.Royal Society of Chemistry.

22. Schoenmakers, P. (2014). Liquid Chromatography: Fundamentals and Instrumentation (2nd ed.). Elsevier.

23. Ahuja, S., & Dong, M.W. (2012). Handbook of Pharmaceutical Analysis by HPLC (Vol. 67).Academic Press.

24. Sajja, R., & Rakesh, K.P. (Eds.). (2018). HPLC Method Development for Pharmaceuticals.CRC Press.

25. Shinkovskaya, Y.V., &Klyushin, D.A. (2017). UHPLC in Pharmaceutical Analysis.Springer.

26. Dolan, J.W., & Snyder, L.R. (2012). Troubleshooting LC Systems: A Comprehensive Approach to Troubleshooting LC Equipment and Separations.John Wiley & Sons.

27. Swartz, M.E., & Krull, I.S. (2012). Modern HPLC for Practicing Scientists.John Wiley & Sons.

28. Dolan, J.W. (2017). Liquid Chromatography: Column Selection Guide. Sigma-Aldrich Technical Bulletin.

29. Cazes, J. (Ed.). (2017). Encyclopedia of Chromatography (3rd ed.). CRC Press.

30. Tarafder, A., Sarkar, M., & Sarkar, B. (2017). Ultra-High Performance Liquid Chromatography: A Short Review. Journal of Chemical and Pharmaceutical Research, 9(4), 15-21.

31. Guillarme, D., &Desmet, G. (2017). Ultra-high performance liquid chromatography: An overview. Journal of Chromatography A, 1523, 1-2.

32. Trabelsi, N., &Borrull, F. (2016). Recent advances in UHPLC columns for environmental analysis. Journal of Separation Science, 39(1), 57-77.

33. Mondello, L., &Tranchida, P.Q. (2017). State-of-the-art in UHPLC: fundamentals, methodologies, and applications in food analysis. Journal of Separation Science, 40(1), 192-202.

34. Saito, Y., &Jinno, K. (2012). Progress in high-performance liquid chromatography and ultra-performance liquid chromatography in Japan. Journal of Chromatography A, 1259, 9-21.

35. Gritti, F., &Guiochon, G. (2013). Theories of peak shape tailing and resolution optimization in gradient elution high-performance liquid chromatography. Journal of Chromatography A, 1313, 107-124.

36. West, C., & Pollard, K. (2017). Advanced practical organic chemistry (3rd ed.). CRC Press.

37. Mazzeo, J.R. (2014). Troubleshooting LC Systems: A Comprehensive Approach to Troubleshooting LC Equipment and Separations.Wiley-VCH.

## Cite this article as:

Gaurav P., Kushwaha S., Tiwari H. and Chandrul K. K. (2024). Review on Application of HPLC and UPLC. *Int. J. of Pharm. & Life Sci.*, 15(3): 33-39.

Source of Support: Nil Conflict of Interest: Not declared For reprints contact: ijplsjournal@gmail.com

International Journal of Pharmacy & Life Sciences